

CONVERSIONS FORMATIVE 1

Solve the problems below, remember that length measurements are dimensional and must be handled differently than mass or volume measurements.

1. $2 \text{ cm} \times 3 \text{ cm} = \underline{\hspace{2cm}}$.


3. $3 \text{ mm} \times 3 \text{ mm} \times 3 \text{ mm} = \underline{\hspace{2cm}}$.

2. $18 \text{ ml} \times 18 \text{ ml} = \underline{\hspace{2cm}}$.


4. $5 \text{ kg} \times 4 \text{ kg} = \underline{\hspace{2cm}}$.

Complete the following conversions, use the picket fence provided.


5. $3 \text{ kg} \rightarrow \text{g}$  =

8. $12,345 \text{ mm} \rightarrow \text{m}$  =

6. $7000 \text{ ml} \rightarrow \text{L}$  =

9. $500 \text{ g} \rightarrow \text{kg}$  =

7. $18\text{m} \rightarrow \text{cm}$ 

10. $0.75\text{L} \rightarrow \text{ml}$ 

Complete the following converts, draw your own picket fence situation.

11. $1,200 \text{ mm} \rightarrow \text{cm}$

14. $0.5 \text{ g} \rightarrow \text{kg}$

12. $0.25 \text{ kg} \rightarrow \text{g}$

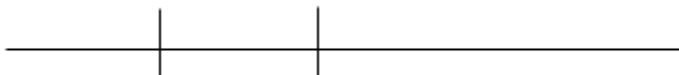
15. $21 \text{ L} \rightarrow \text{mL}$

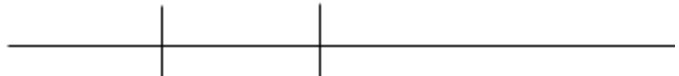
13. $840 \text{ ml} \rightarrow \text{L}$

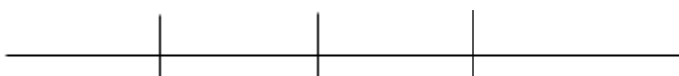
16. $14 \text{ m} \rightarrow \text{mm}$

Density is a property of matter. Density describes how much mass is within a specific unit of volume such as a mL, L, Kg, or cm^3 . Convert the following densities as prompted. Remember $1 \text{ cm}^3 = 1 \text{ cc} = 1 \text{ ml}$.

17. $11,400 \text{ g/L} \rightarrow \text{g/ml}$  =

19. $2.4 \text{ kg/L} \rightarrow \text{g/L}$  =

20. $18.4 \text{ kg/m}^3 \rightarrow \text{kg/cm}^3$  =

21. $0.85 \text{ g/cm}^3 \rightarrow \text{kg/m}^3$  =

22. $0.76 \text{ kg/m}^3 \rightarrow \text{g/mL}$  =